Keywords

Water Treatment; photoelectrocatalysis; Contaminants of emerging concern

Acknowledgements

European Union's Horizon 2020 research and innovation programme under grant agreement No 820718 (PANIWATER)

References

- [1] Tolosana-Moranchel et al., J. Environ. Chem. Eng. 10 (2022) 107955.
- [2] McMichael et al., Appl. Catal. B: Environ. 311 (2022) 121339
- [3] Tolosana-Moranchel et al., Chem. Eng. J. 458 (2023) 141442

3:00 - 3:15pm

O92 2D/2D Heterojunction of NiFe-LDH/MnO $_2$ with promoted visible-light-driven photocatalytic performance for NOx removal

María Ángeles Oliva¹, David Giraldo², Paloma Almodovar², Francisco Martín³, María Luisa López², Ivana Pavlovic¹, <u>Luis Sánchez</u>¹

¹University of Córdoba, Córdoba, Spain. ²Complutense University, Madrid, Spain. ³University of Málaga, Málaga, Spain

Extended Abstract

Summary

In the present study, 2D/2D heterojunctions between ultrathin NiFe-CO₃ layered double hydroxide (LDH) and birnessite (d-MnO₂) were prepared by a self-assembly strategy. The close interface contact between each semiconductor in the NiFe/MnO₂ composite enhanced the photocatalytic performance compared to individual semiconductors. The influence of the atomic ratios on photocatalytic behaviour has been demonstrated, highlighting a 10:1 ratio of NiFe:MnO₂ as the most optimal proportion for a better response. These heterojunctions effectively controlled NOx emissions, being their photooxidative mechanism under visible light here described.

Background

The rapid development of contemporary society has caused a deterioration of air quality [1]. Over the last decades, it has been crucial to devise new remediation strategies to eliminate these pollutants from the atmosphere. In this context, this study proposes the development of the NiFe/MnO₂ composite as a new advanced photocatalyst towards NOx gas removal with enhanced properties, especially strengthening its absorption in the region of the visible light spectrum [2].

Methodology

Given its laminar structure rich in negative charges, birnessite (δ -MnO₂) [3] represents an ideal structural phase for self-assembly with NiFe-LDH nanosheets [4], endowed with positive charges. The structural complementary between delaminated layers of opposite charge facilitates the electrostatic interaction resulting in stable and well-organized composite structures. Finally, several composites were obtained by varying MnO₂ proportions (0, 10, 33 and 66%).

Results and Discussion

Comprehensive characterisation confirmed the successful formation of 2D/2D NiFe/MnO $_2$ composites by a simple self-assembly method through the pure phases of precursors. Figure 1 shows that incorporating a higher MnO $_2$ content in the LDH sample affects textural parameters, reducing specific surface area and pore volume. This decrease limits the number of active centres for anchoring NO gas molecules. Band gap values were determined for NiFe-LDH (2.5 eV) and NiFe/MnO $_2$ composites (1.9 – 2.4 eV range), all able to be activated by visible light. In contrast, the MnO $_2$ (1.7 eV) narrow band gap causes greater light harvesting but supposes higher recombination of charge carriers (e $^-$ /h $^+$).

Photocatalytic DeNOx test demonstrated enhanced performance in NiFe-LDH/MnO $_2$ heterojunctions under visible light, resulting in a notable increase in E_{NO} values and exceptional selectivity for all samples (S = 100%), Figure 2A. The high e^-/h^+ recombination experienced by MnO $_2$ is reduced in NiFe/MnO $_2$ composites, correlating with superior DeNOx activity

results for NiFe/MnO₂-0.1 with E_{NO} = 41 %, against 7% and 23% for MnO₂ and NiFe-LDH, respectively. Figure 2B illustrates a proposed photooxidative mechanism inferred from XPS, EPR and scavenger data. Activated under visible light (λ = 427 nm), the heterojunction promotes the movement of photogenerated electrons to the conduction band (CB) and the creation of holes in the valence band (VB), with an electron flow from NiFe-LDH to MnO₂ CB facilitated by their intimate electronic contact and their edge band position. Additionally, experimental results emphasise the crucial role of superoxide radicals (\cdot O₂⁻¹) in the reaction. The generated radicals subsequently react with environmental molecules, particularly NOx gases, effectively transforming them into nitrate and nitrite species.

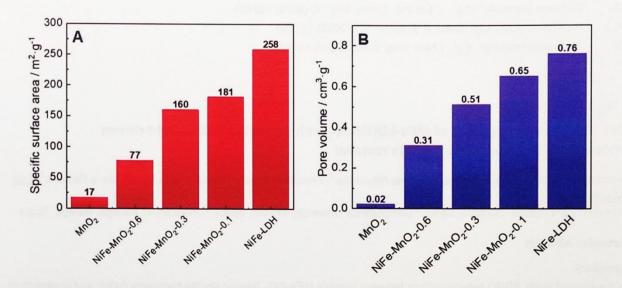


Figure 1. Histogram of the (A) specific surface area and (B) pore volume for the NiFe/MnO₂ system.

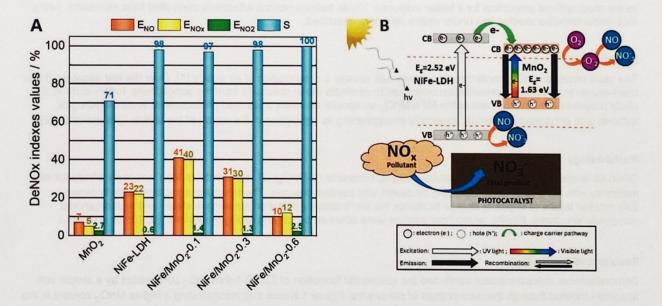


Figure 2. (A) DeNOx indexes of the photocatalytic test under visible light (λ = 427 nm) and (B) schematic diagram for the photodegradation of NOx over NiFe/MnO₂ photocatalysts.

Conclusions

The present study has demonstrated a perfect electronic interaction of a type-I heterostructure. By identifying the optimal ratio of MnO₂ and NiFe-LDH, a favourable band alignment, and establishing correlations in their electronic and physical properties, the recombination of charge carriers was reduced. Furthermore, it optimized the specific surface area and the

formation of reactive oxygen species (ROS). Consequently, these improvements led to a superior photocatalytic performance.

Keywords

LDH; Birnessite; Heterojunction; Photocatalysis; Nitrogen Oxides.

Acknowledgements

This work was financed by the Agencia Estatal de Investigación (MCIN PID2020-117516 GB-I00, PID2020-117832RB-I00), Albufera Energy Storage S.L. (151-2021)(A-2022) and the Spanish Government (FPU19/03570).

References

- [1] Air quality in Europe 2023 report. European Environment Agency, Publications Office of the European Union, Luxembourg.
- [2] Oliva, M. A.et al. Chem. Eng. J., 2023,(470) 144088.
- [3] Drits, V. A.; Lanson, B.; Gaillot. Am. Mineral. 2007, 92 (5-6), 771-788.
- [4] Ma, R.; Liang, J.; Liu, X.; Sasaki, T. J. Am. Chem. Soc. 2012, 134 (48), 19915-19921.